ST. TERESA'S COLLEGE, ERNAKULAM

(AUTONOMOUS)

Affiliated to Mahatma Gandhi University, Kottayam

CURRICULUM FOR
BACHELOR'S PROGRAMME
IN CHEMISTRY

Under Choice Based Credit & Semester System
& Outcome Based Education

(2018 Admissions)

B.Sc. CHEMISTRY

PROGRAMME SPECIFIC OUTCOMES

PSO1: Describe the major concepts and theoretical principles in Chemistry

PSO2: Solve problems using basic understandings in chemistry, physics and mathematics

PSO3: Apply scientific knowledge to design, perform, record and analyze experiments

PSO4: Develop communication skills to identify, investigate, formulate and transmit new ideas and concepts

PSO5: Develop analytical, creative, cognitive skills with social responsibility and environmental consciousness

SEMESTER I

Course Code	Course Title	Credits	Course Type
EN1A01B18	FINE-TUNE YOUR ENGLISH	4	Common Course I
EN1A02B18	PEARLS FROM THE DEEP	3	
MA1A01B18	KATHASAHITHYAM	4	Common Course II
HN1A01B18	KAHAANI AUR UPANYAS	4	
FR1A01B18	FRENCH LANGUAGE AND COMMUNICATIVE SKILLS -I	4	
MT1C01B18	CALCULUS	3	Complementary Course I
PH1C02B18	PROPERTIES OF MATTER & THERMODYNAMICS	2	Complementary Course II
CH1B01B18	GENERAL AND ANALYTICAL CHEMISTRY	2	Core course

SEMESTER I

COMMON COURSE I

EN1A01B18– FINE TUNE YOUR ENGLISH

Credits: 4

Total Lecture Hours: 90

Course Outcomes:

CO1: Recognize the basics of English grammar

CO2: Choose the appropriate word classes

CO3: Identify common errors in the use of English language in various contexts

CO4: Apply the rules of grammar to comprehend, speak, and write grammatically correct English

CO5: Compose materials for business communication

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	3	2
CO3	1	1	1	2	2
CO4	1	1	1	3	2
CO5	1	1	1	2	2

Syllabus Content:

Module 1 **(18 Hours)**

The Sentence and its Structure

How to Write Effective Sentences – Phrases:What are They? – The Noun Clauses – The Adverb Clause – "If All the Trees Were Bread and Cheese" – The Relative Clause – How Clauses are Conjoined

Module 2 **(18 Hours)**

Word-Classes and Related Topics

Understanding the Verb – Understanding Auxiliary Verbs – Understanding Adverbs – Understanding Pronouns – The Reflexive Pronoun – The Articles I – The Articles II – The Adjective – Phrasal Verbs – Mind your Prepositions

Module 3 **(18 Hours)**

To Err is Human

Concord – Errors – Common and Uncommon

Spelling and Pronounciation

Pronunciation: Some Tips – More Tips on Pronunciation – An awesome Mess? – Spelling Part II

Module 4 **(18 Hours)**

Tense and Related Topics

'Presentness' and Present Tenses – The 'Presentness' of a Past Action – Futurity in English – Passivisation

Interrogatives and Negatives

Negatives – How to Frame Questions – What's What? – The Question Tag

Module 5 **(18 Hours)**

Conversational English

Some time expressions – Is John There Please?

Miscellaneous and General Topics

Reading

Letter Writing

In addition there will be an essay question on a general topic.

Learning Resources

Core Text : *Fine-tune Your English* by Dr. Mathew Joseph. Orient Blackswan and Mahatma Gandhi University

SEMESTER I

COMMON COURSE I

EN1A02B18– PEARLS FROM THE DEEP

Credits: 3

Total Lecture Hours: 72

Course Outcomes:

CO1: Name prominent literary figures and recognize various literary devices

CO2: Analyze inherent themes and motives

CO3: Identify the nuances of the age in which the literary work was written

CO4: Examine the different aspects of theatre

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	3	2
CO3	1	1	1	3	2
CO4	1	1	1	3	2

Syllabus Content:

Module 1 (Fiction) **(18hours)**

Ernest Hemingway: The Old Man and the Sea

Module 2 (One Act Plays) **(18hours)**

Susan Glaspell: Trifles

Asif Currimbhoy: The Refugee

A.A Milne: The Boy Comes Home

Module 3 (Short Stories) **(18hours)**

Guy De Maupassant: Two Friends

O. Henry: The Gift of Magi

K.A Abbas: Sparrows

Flora Annie Steel: Valiant Vicky, the Brave Weaver

Module 4 (Poems) **(18hours)**

Rumi: The Chance of Humming

Walter Scott: Lochinvar

John Keats: La Belle Dame Sans Mercy

Robert Frost: After Apple Picking

Chinua Achebe: Refugee Mother and Child

Kamala Das: My Grandmother's House

Ted Hughes: Jaguar

Pablo Neruda: Tonight I can Write the Saddest Lines

P.P Ramachandran: How Simple It Is!

Learning Resources

Core Text: *Pearls from the Deep*. Cambridge University Press and Mahatma Gandhi University

SEMESTER I

COMMON COURSE II

MA1A01B18 – കഥാസാഹിത്യം

Credits: 4

Total Lecture Hours: 72

Course Outcomes:

CO1: ചെറുകഥ, നോവൽ പഠനത്തിലൂടെ വായനാശേഷിയും ആസ്വാദനപ്രാപ്തിയും കൈവരിക്കൽ.

CO2:ചെറുകഥയുടെയും നോവലിന്റെയും കാലാനുസൃതമായ ഭാവുകത്വപരിണാമം തിരിച്ചറിയൽ.

CO3: നിലവിലുള്ളസാമൂഹ്യജീവിത യാഥാർഥ്യങ്ങളെ അഭിമുഖീകരിക്കാൻ പ്രാപ്തരാക്കൽ.

CO4: ആശയവിനിമയം, ഭാഷാവിഷ്കരണം എന്നീ ശേഷികൾ കൈവരിക്കുന്നു

CO5: കഥ,നോവൽ എന്നിവയുടെ വ്യതിരിക്ത സവിശേഷതകൾ തിരിച്ചറിയുന്നു.

CO6:പുതുകാലജീവിതാനുഭവങ്ങൾ വിലയിരുത്താൻ പര്യാപ്തരാകുന്നു

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	2	2
CO3	1	1	1	3	3
CO4	1	1	1	3	2
CO5	1	1	1	2	2
CO6	1	1	1	2	2

Syllabus Content:

ഖണ്ഡംഒന്ന്- 10 മണിക്കൂർ

1.പൂവമ്പഴം -കാരൂർ

2.ഭൂമിയുടെഅവകാശികൾ -വൈക്കംമുഹമ്മദ്ബഷീർ

ഖണ്ഡംരണ്ട് - 15മണിക്കൂർ

1.കടൽ -ടി .പദ്മനാഭൻ

2.പെരുമഴയുടെപിറ്റേന്ന് -എം. ടി. വാസുദേവൻനായർ

3.മാനാഞ്ചിറടെസ്റ്റ് -വി .കെ.എൻ

4.തരിശുനിലം -മാധവിക്കുട്ടി

ഖണ്ഡംമൂന്ന് - 15മണിക്കൂർ

1.ആർക്കറിയാം -സക്കറിയ

2.ഓരോഎഴുത്തുകാരിയുടെഉള്ളിലും -സാറാജോസഫ്

3.തിരുത്ത് -എൻ .എസ് .മാധവൻ

4.മോഹമഞ്ഞ -കെ .ആർ .മീര

ഖണ്ഡംനാല്-10 മണിക്കൂർ

1.അഗ്നി -സിതാര.എസ്

2.ബിരിയാണി -സന്തോഷ്എച്ചിക്കാനം

3.മോദസ്ഥിരനായി അങ്ങ് വസിപ്പൂമലപോലെ -എസ്. ഹരീഷ്

4.സ്നേഹബഹുമാനപെട്ടഅന്നാമ്മയ്ക്ക്ഗീതാലക്ഷ്മിഎഴുതുന്നകത്ത് -പ്രിയഎ .എസ്

5.ചിലസ്വപ്നങ്ങളിൽ .....സീതാലക്ഷ്മിയുടെകറുത്തമുടിയിഴ -ഇന്ദുമേനോൻ

ഖണ്ഡംഅഞ്ച് - 22മണിക്കൂർ

ആടുജീവിതം -ബന്യാമിൻ

സഹായകഗ്രന്ഥങ്ങൾ

1.ചെറുകഥഇന്നലെഇന്ന് -എം .അച്യുതൻ

2.ചെറുകഥാപ്രസ്ഥാനം -എം.പി. പോൾ

3.ചെറുകഥ വാക്കുംവഴിയും -ഡോ .കെ.എസ്.രവികുമാർ

4.നോവൽസാഹിത്യചരിത്രം -പ്രൊഫ.കെ.എം .തരകൻ

SEMESTER I

COMMON COURSE II

HN1A01B18- KAHAANI AUR UPANYAS

Credits: 4

Total Lecture Hours: 72

Course Outcomes:

CO1: Discuss story content and structure in depth.

CO2: Analyse characterisation and comment on the development of the characters as the story/ novel unfolds.

CO3: Analyse short stories and novels on the basis of literary elements like plot, theme, metaphor, and image.

CO4: Compare treatments of theme, character and subject matter of different short stories.

CO5: Illustrate greater reading fluency and improved vocabulary in Hindi.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	2	2
CO3	1	1	1	2	2
CO4	1	1	1	2	2
CO5	1	1	1	3	2

Syllabus Content:

Module I: (18 Hrs)

ANTHIM SAAKSHYA –CHANDRAKAANTA CHAPTERS 1 ,2

EIDGAAH- PREMCHAND

Module II: (20 hrs)

ANTHIM SAAKSHYA –CHANDRAKAANTA CHAPTERS 3, 4, 5 JANGAL KA DAAH- SWAYAM PRAKASH CHCHUTTI KA DIN- USHA PRIYAMVADA

Module III: (20 hrs)

ANTHIM SAAKSHYA –CHANDRAKAANTA CHAPTERS 6,7,8 MAA RASOI MEI REHTI HAI – KUMAR AMBUJ KHEER – MADHAVI KUTTY

Module IV: (16 hrs)

ANTHIM SAAKSHYA –CHANDRAKAANTA CHAPTERS 9, 10 HEELIBON KI BATHTHAKHE- AGYEY

SEMESTER I

COMMON COURSE II

FR1A01B18- FRENCH LANGUAGE AND COMMUNICATIVE SKILLS -I

Credits: 4

Total Lecture Hours: 72

Course Outcomes:

CO1: Describe topics such as family, professions, time, place, likes and dislikes, daily life situations.

CO2: Develop language, vocabulary and grammar skills.

CO3: Articulate various speech sounds and their determined combinations.

CO4: Prepare conversations based on scenarios which helps while traveling

CO5: Articulate the concepts to express one's opinion in a specific situation.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	3	2
CO3	1	1	1	3	2
CO4	1	1	1	2	2
CO5	1	1	1	3	2

Syllabus Content:

Module I : (25 hours)

La population L"alphabet – Les chiffres – Identité – Se présenter – Poser des questions – Les professions – Les nationalités

Module II : (23 hours)

La banlieue Demander une information, un prix – l"heure – la ville

Module III : (24 hours)

Quartier de Paris Décrire un lieu – Indiquer un prix, un itinéraire.

SEMESTER I

COMPLEMENTARY COURSE I

MT1C01B18– DIFFERENTIAL AND INTEGRAL CALCULUS

Credits: 3

Total Lecture Hours: 72

CO1: Evaluate the rate of change of functions using the definition of limit and the differentiation rules.

CO2: Apply the concept of differentiation to find the extreme values of a function and interpret the consequences of Rolle's theorem and Mean value theorem for differentiable functions.

CO3: Interpret the area under the curve as a definite integral and find the area between curves.

CO4: Apply integration to calculate lengths of plane curves, areas of surfaces of revolution and volumes by slicing and rotation about an axis.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	2	1	1	2
CO2	1	3	1	1	2
CO3	1	3	2	1	2
CO4	1	3	1	1	2

Syllabus Content:

Module 1 :Differential Calculus: (22 Hrs)

Rates of change and limits, calculating limits using the limit laws, the precise definition of a limit, one sided limits and limits at infinity, derivative of a function, differentiation rules, the derivative as a rate of change, derivatives of trigonometric functions, the chain rule and parametric equations, implicit differentiation.

Module 2: Applications of Derivatives: (15 Hrs)

Extreme values of functions, The Mean Value Theorem, Monotonic functions and the first derivative test.

Module 3 : Integral Calculus: (15 Hrs)

A quick review of indefinite integral as anti-derivative,The Definite integral, The fundamental theorem of Calculus

Module 4: Application of Integrals: (20Hrs)

Substitution and area between curves, Volumes by slicing and rotation about an axis (disc method only), Lengths of plane curves, Areas of surfaces of revolution and the theorem of Pappus (excluding theorem of Pappus).

SEMESTER I

COMPLEMENTARY COURSE I

PH1C02B18: PROPERTIES OF MATTER & THERMODYNAMICS

Credits: 2

Total Lecture Hours: 36

CO1: Apply static and dynamic methods to determine rigidity modulus and bending of beams to Young's modulus

CO2: Discuss the theory for the dynamics of fluid systems

CO3: Examine Carnot engine and refrigerator by applying second law of thermodynamics

CO4: Deduce Maxwell's thermodynamic relations from thermodynamic potentials

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	3	2	1	2
CO2	1	3	2	1	2
CO3	2	3	2	1	2
CO4	2	3	1	1	2

Syllabus Content:

Module I: Elasticity (13 hours)

Stress- strain- Hooke's law- Elastic moduli- Poisson's ratio- twisting couple determination of rigidity modulus- static and dynamic methods- static torsion- torsion pendulum, bending of beams- cantilever, uniform and non-uniform bending, I section girder.

Text Book :*Mechanics - D. S. Mathur- Revised by P. S. Hemne, S. Chand & Co., Chapters 13 & 14.*

Module II: Surface tension (3 hours)

Molecular theory of surface tension - surface energy - excess pressure in a liquid drop, factors affecting surface tension – applications

Text book:*Mechanics– Prof. D.S Mathur Revised by: Dr. P.S Hemne. , S Chand & Company Pvt. Ltd, chapter 15*

Hydrodynamics (7 hours)

Streamline and turbulent flow - critical velocity - Coefficient of viscosity - Derivation of Poiseuille's equation, Stokes equation-Determination of viscosity by Poiseuille's method-Brownian motion – Viscosity of gases – Bernoulli's theorem.

Text book: ***Properties of Matter- Brijlal and N. Subrahmaniam, S. Chand &Company Pvt. Ltd,1989, Chapter 8***

Module III: Thermodynamics (13 hours)

Thermodynamic systems- thermodynamic equilibrium- thermodynamic processes isothermal process- adiabatic process- zeroth law of thermodynamics, first law of thermodynamics- heat engine- the Carnot engine- refrigerator, concept of entropy second law of thermodynamics- third law of thermodynamics- Maxwell's thermodynamic relations.

Text Book ***: Heat and Thermodynamics, Brijlal and Subrahmanyam and P. S. Hemne, S. Chand & Co., Chapter 5 & 6***

SEMESTER I

CORE COURSE

CH1B01B18: GENERAL AND ANALYTICAL CHEMISTRY

Credits: 2

Total Lecture Hours: 36

Course Outcomes:

CO1: Compute the number of significant digits, mean and standard deviation, percentage and distribution of errors from a set of analytical data and the molecular mass, molarity, oxidation and reduction numbers, equivalent mass.

CO2: Explain the methodology of chemistry and the periodic properties of elements

CO3: Illustrate the principles of analytical chemistry in the intergroup separation of cations, methods of expressing concentration, quantification of analytes by titrimetry, gravimetry and separation of organic compounds

CO4: Differentiate between column chromatography, TLC, GC, Ion exchange chromatography and HPLC

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	2	1	2
CO2	3	2	2	1	1
CO3	3	2	3	1	2
CO4	3	2	2	1	1

Syllabus Content:

Module 1: Methodology of Chemistry and Evaluation of Analytical Data(12Hrs)

Definition of Science. Scientific methods - observation-posing a question - formulation of hypothesis- experiment – theory - law. Falsification of hypothesis - inductive and deductive reasoning- revision of scientific theories and laws.

Evolution of Chemistry-ancient speculation on the nature of matter. Early form of chemistry-alchemy, origin of modern chemistry. Structure of chemical science: Scope, theory and experiment - branches of chemistry. Role of chemistry as a central science connecting physics, biology and other branches of science. Interdisciplinary areas involving chemistry: Nanotechnology and biotechnology.

Evaluation of Analytical Data: Units, significant digits, rounding, scientific and prefix notation, graphing of data. Precision and accuracy-types of errors – ways of expressing precision – ways to reduce systematic errors - reporting analytical data. Statistical treatment of analytical data – population and samples –Mean and standard deviation – distribution of random errors.

Module II: Periodic Table and Properties (6 Hrs)

Modern periodic law – Long form periodic table. Diagonal relationship and anomalous behavior of first element in a group. Periodicity in properties: Atomic and ionic radii - ionization enthalpy - electron affinity (electron gain enthalpy) – electronegativity. Electronegativity scales: Pauling and Mullikan scales. Effective nuclear charge – Slater rule and its applications – polarising power.

Molecular mass - mole concept – molar volume. Oxidation and reduction – oxidation number and valency – variable valency - equivalent mass of oxidizing agent and reducing agent using oxidation number concept.

Module III : Analytical Methods in Chemistry (18 Hrs)

Qualitative analysis: Applications of solubility product and common ion effect in the precipitation of cations. Principle of intergroup separation of cations. Interfering acid radicals and their elimination (oxalate, fluoride, borate and phosphate).

Titrimetric analysis - fundamental concepts. Methods of expressing concentration: Weight percentage, molality, molarity, normality, mole fraction, ppm. and ppb. Primary and secondary standards, quantitative dilution – problems. Acid base titrations- titration curves – pH indicators. Redox titrations – titration curve –titrations involving MnO_4^- and $Cr_2O_7^{2-}$ - redox indicators. Complexometric titrations – EDTA titrations - titration curves – metal ion indicators. Gravimetric analysis: Unit operations in gravimetric analysis - illustrations using iron and barium estimation.

Separation and purification techniques – filtration, crystallization and precipitation – fractional distillation, solvent extraction.

Chromatographic Methods: Column Chromatography: Principle, types of adsorbents, preparation of the column, elution, recovery of substances and applications. Thin Layer Chromatography: Principle, choice of adsorbent and solvent, preparation of Chromatoplates, Rf-values, significance of Rf values. Ion exchange chromatography: Principle and experimental techniques. Gas Chromatography: Principle and experimental techniques. High Performance Liquid Chromatography (HPLC): Principle and experimental techniques.

SEMESTER II

Course Code	Course Title	Credits	Course Type
EN2A03B18	ISSUES THAT MATTER	4	Common Course I
EN2A04B18	SAVOURING THE CLASSICS	3	
MA2A03B18	KAVITHA	4	Common Course II
HN2A03B18	KAVITHA VYAKARAN AUR ANUVAD	4	
FR2A03B18	FRENCH LANGUAGE AND COMMUNICATIVE SKILLS -II	4	
MT2C01B18	PARTIAL DERIVATIVES, MULTIPLE INTEGRALS TRIGONOMETRY AND MATRICES	3	Complementary Course I
PH2C02B18	MECHANICS AND CRYSTALLOGRAPHY	2	Complementary Course II
PH2CP02B18	PRACTICAL	2	Complementary Practical
CH2B02B18	THEORETICAL AND INORGANIC CHEMISTRY	2	Core course
CH2BP01B18	VOLUMETRIC ANALYSIS	2	Core Practical

SEMESTER II

COMMON COURSE I

EN2A03B18 - ISSUES THAT MATTER

Credits: 4

Total Lecture Hours: 90

Course Outcomes:

CO1. Identify the major issues of contemporary significance

CO2. Discuss the consequences of war and refugee crisis with respect to the psychological dimension

CO3. Employ theoretical learning in classrooms to current developments in the world

CO4. Critique the diverse experiences both historical and contemporary to create a more informed vision of the future

CO5. Develop oneself as a conscious, concerned, conscientious human being

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	3	2
CO3	1	1	1	1	2
CO4	1	1	1	2	2
CO5	1	1	1	1	2

Syllabus Content

Module 1 **(18 hours)**

"The Unsurrendered People" – Kenzaburo Oe

"The Old Prison" – Judith Wright

"War" – Luigi Pirandello

Module 2 **(18 hours)**

Persuasions on the Power of the Word:

"On Censorship" – Salman Rushdie

"Peril" – Toni Morrison

"The Burning of the Books" – Bertolt Brecht

"The Censors" – Luisa Valenzuela

Module 3 **(18 hours)**

"The Poisoned Bread" – Bandhu Madhav

"A Trip Westward" – Zitkala-Sa

"The Pot Maker" – Temsula Ao

Module 4 **(18 hours)**

"Does it Matter?" – Richard Leakey

"On Killing a Tree" – Gieve Patel

"Hagar: A Story of a Woman and Water" (Gift in Green (chapter 2)) – Sarah Joseph

Module 5 **(18 hours)**

"Understanding Refugeeism: An Introduction to Tibetan Refugees in India" – Mallica Mishra

"Refugee Blues" – W.H Auden

"The Child Goes to the Camp" (from Palestine's Children) – Ghassan Kanafani

SEMESTER II

COMMON COURSE I

EN2A04B18 – SAVOURING THE CLASSICS

Credits: 3

Total Lecture Hours: 72

Course Outcomes:

CO1: Recognise the time-tested literary masterpieces from diverse cultures

CO2: Identify the representative authors from various genres (poetry, drama, novel, short fiction)

CO3: Recite celebrated lines from Classic works

CO4: Discuss the 'universals' of human condition

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	2
CO2	1	1	1	1	2
CO3	1	1	1	3	2
CO4	1	1	1	3	2

Syllabus Content

Module 1 (Poems) **(18hours)**

Homer: "Father and Son" (Odyssey Book 16: 113-189) (Translated by Robert Fagles)

Kalidasa: "Lovely is Youth" (Translated by J.G Jennings)

Omar Khayyam: Rubaiyat (quatrains: 25-28) (Translated by Edward Fitzgerald)

Dante: Dante meets Virgil (Inferno Canto 1: 49-102) (Translated by J.G Nichols)

John Milton: "On his Blindness"

Module 2 (Shakespeare Excerpts) (18hours)

Romeo and Juliet: Act II, Scene ii

The Merchant of Venice: Act IV, Scene i

Module 3 (Novel Excerpts) (18hours)

Miguel de Cervantes: Don Quixote (Chapter 8) (Translated by Edith Grossman)

Jane Austen: Pride and Prejudice (Chapters 1-6)

Victor Hugo: Les Miserables (Part 1- Fantine, Book II, Chapters 9-13) (Translated by Christine Donougher)

Module 4 (Short Fiction) (18hours)

Charles Dickens: The Black Veil

Leo Tolstoy: How Much Land Does a Man Need? (Translated by Louise & Aulmer Maude)

Rabindranath Tagore: Kabuliwala (Translated by Mohammad A Quayum)

Jorge Louis Borges: The Shape of the Sword (Translated by Andrew Hurley)

സെമസ്റ്റർ രണ്ട്

കോമൺ കോഴ്സ് II- മലയാളം

MA2A03B18-കവിത

ക്രെഡിറ്റ് : 4

പഠനസമയം : 72 മണിക്കൂർ

കോഴ്സ് ഔട്ട്കം (Course Outcome)

CO1:പത്തൊൻപത് കവിതകളുടെ പഠനത്തിലൂടെ വായനാശേഷിയും ആസ്വാദന പ്രാപ്തിയും കൈവരിക്കൽ.

CO2:മലയാളകവിതകളിലെ കാലാനുസൃതമായ ഭാവുകത്വപരിണാമം തിരിച്ചറിയ.

CO3:നിലവിലുള്ള സാമൂഹ്യജീവിതയാഥാർഥ്യങ്ങളെ അഭിമുഖീകരിക്കാൻ പ്രാപ്തരാക്കൽ.

CO4:പരിസ്ഥിതികസൗന്ദര്യശാസ്ത്രത്തെയും ചില സാമൂഹ്യചരിത്ര പശ്ചാത്തലങ്ങളെയും കുറിച്ച് ഗ്രഹിക്കൽ.

CO5:വിദ്യാർത്ഥികളുടെ സർഗ്ഗാത്മകശേഷി വികസിക്കൽ .

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	3
CO2	1	1	1	2	3
CO3	1	1	1	1	3
CO4	1	1	1	1	3
CO5A	1	1	1	2	3

ഖണ്ഡം ഒന്ന്- 20 മണിക്കൂർ

1. മാംസനിബദ്ധമല്ല രാഗം -കുമാരനാശാൻ (ലീലയിലെ 47 മുതൽ 74 വരെയുള്ള 28 ശ്ലോകങ്ങൾ)

2.സ്നേഹസുന്ദരപാതയിലൂടെ -വൈലോപ്പിള്ളി ('കുടിയൊഴിക്കലി'ലെ അവസാന ഖണ്ഡം)

ഖണ്ഡം രണ്ട് 15 മണിക്കൂർ

1 .ഒറ്റയ്ക്കിരിക്കാൻ പഠിച്ചുകഴിഞ്ഞു ഞാൻ -സുഗതകുമാരി

2 .കോഴി -കടമ്മനിട്ടരാമകൃഷ്ണപിള്ള

3 .പഴഞ്ചൊല്ലുകൾ -സച്ചിദാനന്ദൻ

4 .മുള്ളൻപന്നി -കെ.ജി.ശങ്കരപ്പിള്ള

ഖണ്ഡം മൂന്ന് 15 മണിക്കൂർ

1.തിരുത്ത്-പി .പി.രാമചന്ദ്രൻ

2.പിറക്കാത്ത മകന് -ബാലചന്ദ്രൻ ചുള്ളിക്കാട്

3.മൃഗശിക്ഷകൻ -വിജയലക്ഷ്മി

4.കുന്നിമണികൾ-കുഞ്ഞുണ്ണി

ഖണ്ഡം നാല് 22 മണിക്കൂർ

1 .ആടിയാടില അലഞ്ഞ മരങ്ങളേ -അൻവർ അലി

2 .കൽവീട് -വി.എം.ഗിരിജ

3 . ആഴങ്ങൾ അടച്ചിട്ട പുഴ -എസ് .ജോസഫ്

4 .സ്മാരകം -വീരാൻകുട്ടി

5 .കുട്ടമ്മാൻ -എം.ർ.രേണുകുമാർ

6 .നാഷണൽ ജ്യോഗ്രഫി -എസ് .കണ്ണൻ

7 .വാഴക്കുല -കെ .ആർ.ടോണി

8 .പഴയ ചിലത് -പി.രാമൻ

9 .ഗോതമ്പുശില്പം -കവിത ബാലകൃഷ്ണൻ

SEMESTER II

COMMON COURSE II

HN2AO3B18 - KAVITA, VYAKARAN AUR ANUVAD

Credits – 4

Total Hours- 72

Course Outcomes:

CO1: Contextualize and summarise the poems of different genres in Hindi

CO2: Evaluate the Poets contribution to Hindi literature

CO3: Demonstrate linguistic ability for translation of texts between Hindi & English

CO4: Classify Parts of Speech

CO5: Illustrate greater fluency in Hindi by applying theoretical knowledge of Grammar

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	3	3
CO2	1	1	1	2	3
CO3	1	1	1	3	2
CO4	1	1	1	3	3
CO5	1	1	1	3	2

Syllabus Contents

Module I **(18 Hours)**

Vyaakaran

Module II **(20 Hours)**

Tulasidas

Kabir

Ve Muskathe Phool Nahi- Mahadevi Verma

Cheenane Aaye Hain Ve – Sarweshvar Dayal Saxena

Dilli Darwaaza – Kumar Vimal

Jungle Ke Ujaad Mei – Vinod Kumar Shukla

Aazadi Urf Gulaami – Gyanendrapathi

Module III **(20 Hours)**

Meera

Bazaar- Mangalesh Dabraal

Beesvi Sadi Ke Antim Dino Ka Aashcharya- Rajesh Joshi

Do Haathiyon Ki Ladaai- Uda Pakash

Thande Paani Ki Machine – Ekant Srivastav

Saboot – Arun Kamal

Tumhe Kuch Karna Chahiye – Chanrakanth Devthale

Module IV **(14 Hours)**

Anuvaad

SEMESTER II

COMMON COURSE II

FR2A03B18- FRENCH LANGUAGE AND COMMUNICATIVE SKILLS -II

Credits: 4

Total Lecture Hours: 72

Course Outcomes:

CO1: Identify familiar everyday expressions and basic phrases.

CO2: Ask questions to get meaningful responses in effective communication.

CO3: Develop language, vocabulary and grammar skills.

CO4: Prepare conversations based on various situations

CO5: Articulate the concepts to express one's opinion in a specific situation.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	3	3
CO2	1	1	1	3	3
CO3	1	1	1	3	3
CO4	1	1	1	3	3
CO5	1	1	1	3	3

Syllabus Content:

Module I **(25 hours)**

Chambre pour étudiants Localiser des objets – l'habitat – les meubles – l'appréciation

Module II **(23 hours)**

Petits boulots Téléphoner – Raconter – l'emploi

Module III **(24 hours)**

Le resto U Exprimer une opinion – Poser des questions – la nourriture

SEMESTER II

COMPLEMENTARY COURSE I

MT2C01B18 - PARTIAL DERIVATIVES, MULTIPLE INTEGRALS, TRIGONOMETRY AND MATRICES

Credits: 2

Total Lecture Hours: 72

Course Outcomes

CO1 : Apply the multiple integrals to calculate quantities that vary over two and three dimensions.

CO2 : Illustrate the separation of Circular and hyperbolic functions of a complex variable function into real and imaginary parts and expansions of some trigonometric functions and employ C + iS method for the summation of an infinite series

CO3 : Calculate the Partial derivatives using the rules of differentiation. .

CO4 : Explain the properties of Matrices and employ the Matrix Algebra to solve a system of linear equations , determine the characteristic roots and characteristic vectors of a Matrix and illustrate the Cayley Hamilton theorem.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	3	1	1	2
CO2	1	3	1	1	2
CO3	1	3	1	1	2
CO4	1	3	1	1	2

Syllabus Content:

Module 1 Multiple Integrals: (17 Hours)

Double Integrals, area of bounded region in plane only, Double Integrals in Polar form, Triple integrals in rectangular co-ordinates, Volume of a region in space (As in Sections 15.1, 15.2, 15.3, 15.4 of Text 1)

Module 2 Trigonometry: (20Hours)

Expansions of functions like $\sin n\theta$, $\cos n\theta$, $\tan n\theta$, $\cos^n\theta$, hyperbolic functions, inverse circular and hyperbolic function. Separation into real and imaginary parts. Summation of infinite series based on C+iS method. (Geometric, Binomial, Exponential, Logarithmic and Trigonometric series)

(Relevant Sections in Chapter 3 to 5 and Chapter 8of Text 3)

Module 3 Partial Derivatives: (15 Hours)

Functions of several variables (Definition only), Partial derivatives, The Chain Rule (Sections 14.3 - 14.4 of Text 1)

Module 4 Matrices: (20Hours)

Rank of a Matrix, Non-Singular and Singular matrices, Elementary Transformations, Inverse of an elementary Transformations, Equivalent matrices, Row Canonical form, Normal form, Elementary matrices only.

Systems of Linear equations: System of non-homogeneous, solution using matrices, Cramer's rule, system of homogeneous equations, Characteristic equation of a matrix; Characteristic roots and characteristic vectors. Cayley-Hamilton theorem (statement only) and simple applications

(Chapters – 5, 10, 19, 23 of text 2).

SEMESTER II

COMPLEMENTARY COURSE II

PH2C02B18- MECHANICS AND CRYSTALLOGRAPHY

Credits - 2

Total lecture hours - 36 hrs

Course Outcomes:

CO1: Articulate the motion under gravity and determine the acceleration due to gravity

CO2: Apply relevant theorems and strategies to determine the physical parameters related to rotational motion of bodies

CO3: Represent and solve equations of oscillatory motion of particles

CO4: Summarize different crystal systems and X-ray diffraction in crystals

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	3	3	1	2
CO2	1	3	1	1	2
CO3	1	3	1	1	2
CO4	1	3	3	1	2

Syllabus Content

Module I Motion under gravity (5 hours)

Velocity- acceleration- force – acceleration due to gravity - compound pendulum (symmetric and asymmetric) radius of gyration –centripetal acceleration and force - centrifugal force

Rotational dynamics (10 hours)

Angular velocity- angular momentum- torque- conservation of angular momentum angular acceleration- moment of inertia- parallel and perpendicular axes theorems moment of inertia of rod, ring, disc, cylinder and sphere- flywheel

Module II

Oscillations (9 hours)

Periodic and oscillatory motion- simple harmonic motion- differential equation, expression for displacement, velocity and acceleration- graphical representation- energy of a particle executing simple harmonic motion damped oscillation- forced oscillation and resonance.

Waves (4 hours)

Waves-classifications- progressive wave- energy of progressive wave- superposition of waves- theory of beats- Doppler effect.

Module III

Crystalline Solids (8 hours)

Crystalline and amorphous solids – crystal lattice and translation vectors – basis– unit cell – lattice parameters – crystal systems – crystal planes and directions –Miller indices – inter planar spacing – hcp, fcc, bcc, sc crystal structures –Bragg's law of X ray diffraction.

SEMESTER II

PRACTICAL

PH2CP02B18 - MECHANICS AND CRYSTALLOGRAPHY

Credit - 2

No. of hours: 72

Course Outcomes:

CO1: Apply the knowledge of basic concepts in Physics to identify and select appropriate measuring instruments.

CO2: Analyse basics experiments in Properties of Matter, Mechanics and construct diode circuits.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	3	3	1	2
CO2	1	3	3	1	2

Syllabus Content

1. Vernier Calipers - Volume of a cylinder- sphere and a beaker
2. Screw gauge - Volume of a sphere and a glass plate
3. Beam balance - Mass of a solid (sensibility method)
4. Radius of a capillary tube- Using (1) travelling microscope
5. Density of a liquid - U-Tube and Hare's apparatus
6. Viscosity of a liquid - Variable pressure head
7. Surface Tension – Capillary rise method.
8. Cantilever - Pin & Microscope – Determination of Young's Modulus
9. Symmetric Compound Pendulum-Determination of radius of gyration(K) and Acceleration due to gravity (g)

10. Spectrometer – Angle of the Prism.
11. Cantilever – Scale and Telescope-Determination of Young's modulus
12. Asymmetric Compound Pendulum-Determination of K and g
13. Coefficient of Viscosity – Constant pressure head
14. Spectrometer - Refractive Index of material of prism.
15. Liquid lens - Refractive Index of glass using liquid of known refractive index
16. Potentiometer-Calibration of low range voltmeter
17. Characteristics of Zener diode
18. Construction of half wave rectifier with and without filter – Ripple factor and Load regulation
19. Characteristics of p-n junction diode
20. Torsion pendulum - Rigidity modulus

SEMESTER - II

CORE COURSE

CH2B02B18: THEORETICAL AND INORGANIC CHEMISTRY

Credits - 2

Total Lecture Hours : 36

Course Outcomes

CO1: Apply the knowledge of atomic theories in recognizing the atomic spectra, wave particle dualities, Heisenberg's uncertainty principle and quantum numbers.

CO2: Explain the properties of s and p block elements, transition metals and lanthanides.

CO3: Explain the properties of ionic bond and ionic solids, calculation of lattice energy of ionic solids from Born-Lande equation and Born-Haber cycle, features of intermolecular forces and theories of metallic bond.

CO4: Illustrate the covalent bonding in molecules using Valence Bond Theory, Resonance concept, Hybridization, VSEPR theory and Molecular Orbital theory.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	2	1	1
CO2	3	2	2	1	1
CO3	3	2	2	1	1
CO4	3	3	2	1	1

Syllabus Content

Module 1: Atomic Structure **(6 Hrs)**

Introduction based on historical development (Dalton's atomic theory, Thomson's atom model Rutherford's atom model) - failure of classical physics – black body radiation - Planck's quantum hypothesis - photoelectric effect - generalization of quantum theory . Atomic spectra of hydrogen

and hydrogen like atoms– Bohr theory of atom – Calculation of Bohr radius, velocity and energy of an electron - explanation of atomic spectra - limitations of Bohr theory. Louis de Broglie's matter waves – wave-particle duality - electron diffraction - Heisenberg's uncertainty principle. Schrödinger wave equation (derivation not expected), wave functions – significance of ψ and ψ2 – atomic orbitals and concept of quantum numbers - shapes of orbitals (s, p and d) - Pauli's Exclusion principle - Hund's rule of maximum multiplicity - Aufbau principle – electronic configuration of atoms.

Module II : Chemistry of s, p, d & f Block Elements (10 Hrs)

s and p block: Periodicity in s-and p- block elements with respect to electronic configuration, atomic and ionic size, ionization energy and electro negativity. Inert pair effect.

Transition Metals: General characteristics: Metallic character, oxidation states, size, density, melting points, boiling points, ionization energy, colour, magnetic properties, reducing properties, catalytic properties, non-stoichiometric compounds, complex formation and alloy formation. Difference between first row and other two rows. Preparation, properties, structure and uses of KMnO4 and K2Cr2O7.

Lanthanides: Electronic configuration and general characteristics – Occurrence of lanthanides. Isolation of lanthanides from monazite sand - Separation by ion exchange method. Lanthanide contraction: Causes and consequences. Industrial importance of lanthanides.

Module III : Chemical Bonding (20 Hrs)

Introduction – Octet rule and its limitations. Types of bonds: Ionic bond - factors favouring the formation of ionic bonds - lattice energy of ionic compounds - Born- Lande equation with derivation - solvation enthalpy and solubility of ionic compounds – Born-Haber cycle and its applications – properties of ionic compounds - polarisation of ions – Fajan's rule and its applications.

Covalent Bond: Valence Bond Theory and its limitations. Concept of resonance - resonance

structures of borate, carbonate and nitrate ions. Hybridization: Definition and characteristics

– shape of molecules (BeCl2, C2H2, BF3, C2H4, CH4, PCl5, SF6 and IF7). VSEPR theory: Postulates - applications - shapes of molecules NH3, H2O, XeF2, IF5, and XeF6.

Properties of covalent compounds - polarity of bonds – percentage of ionic character – dipole moment and molecular structure.

Covalent Bond: Molecular Orbital Theory – LCAO - bonding and anti-bonding molecular orbitals – bond order and its significance. MO diagrams of homonuclear and heteronuclear diatomic molecules: H2, He2, Li2, Be2, B2, C2, N2, O2, F2, CO and NO – comparison of bond length, magnetic behavior and bond energy of O2, O2+, O22+, O2- and O22- . Metallic

Bond: free electron theory, valence bond theory and band theory (qualitative treatment only) - explanation of metallic properties based on these theories.

Intermolecular forces: Hydrogen bond - intra and inter molecular hydrogen bonds – effect on physical properties. Van der Waals forces, ion-dipole, dipole-dipole, ion-induced dipole, dipole-induced dipole and induced dipole-induced dipole interactions

SEMESTER I AND II
CORE CHEMISTRY PRACTICALS

CH2BP01B18 - VOLUMETRIC ANALYSIS

Credits: 2

Total Hours: 72

Course Outcomes:

CO1: Prepare standard solutions for microscale volumetric analysis.

CO2: Record the molarity of the given intermediate solution by standardizing it.

CO3: Calculate the mass of the analyte in a given solution by microscale volumetric analysis.

CO4: Administer microscale analysis of solutions by different types of volumetry like acidimetry, alkalimetry, complexometry, permanganometry, dichrometry, iodometry and iodimetry.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	2	3	3	1	2
CO2	2	3	3	1	2
CO3	2	3	3	1	2
CO4	2	3	3	1	2

Syllabus Content:

Micro Analysis

A. Acidimetry and Alkalimetry

1. Strong acid-Strong base
2. Strong acid – Weak base
3. Strong base – Weak acid
4. Estimation of $Na_2CO_3$and $NaHCO_3$ in a mixture
5. Estimation of NaOH and Na_2CO_3 in a mixture
6. Estimation of ammonia in ammonium salts by direct and indirect methods

B. Complexometric Titrations Using EDTA

1. Estimation of Zn
2. Estimation of Mg
3. Estimation of Mg and Ca in a mixture
4. Estimation of Ni
5. Determination of hardness of water

C. Oxidation – Reduction Titrations

(i) Permanganometry

1. Estimation of ferrous iron
2. Estimation of oxalic acid
3. Estimation of sodium oxalate
4. Estimation of calcium

(ii) Dichrometry

1. Estimation of ferrous iron using internal indicator
2. Estimation of ferrous iron using external indicator
3. Estimation of ferric iron using internal indicator
4. Estimation of ferric iron using external indicator

(iii) Iodimetry and Iodometry

1. Estimation of copper
2. Estimation of arsenious oxide

SEMESTER III

Course Code	Course Title	Credits	Course Type
EN3A05B18	LITERATURE AND/AS IDENTITY	4	Common Course I
MA3A05B18	DRISYAKALASAHITHYAM	4	Common Course II
HN3A05B18	NAATAK AUR LAMBI KAVITA	4	
FR3A05B18	AN ADVANCED COURSE IN FRENCH –I	4	
MT3C01B18	VECTOR CALCULUS, DIFFERENTIAL EQUATIONS AND ANALYTIC GEOMETRY	4	Complementary Course I
PH3C02B18	MODERN PHYSICS AND BASIC ELECTRONICS	3	Complementary Course II
CH3B03B18	ORGANIC CHEMISTRY – I	3	Core course

SEMESTER III

COMMON COURSE I

EN3A05B18 – LITERATURE AND/AS IDENTITY

Credits: 4

Total Lecture Hours: 90

Course Outcomes:

CO1. Explain how literature problematizes identity.

CO2. Analyze the quest for identity in the Indian diaspora.

CO3. Illustrate the effects of partition and communal violence in South Asian Literature.

CO4. Critique the social construction of identity.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	3
CO2	1	1	1	2	3
CO3	1	1	1	2	3
CO4	1	1	1	2	3

Syllabus Content

Module 1 (Diasporic Identities) **(18 hours)**

Agha Shahid Ali: Postcard from Kashmir

Amy Tan: Mother Tongue

Imtiaz Dharker: At the Lahore Karhai

Chitra Banerjee Divakaruni: Indian Movie, New Jersey

Module 2 (South Asian Identities) **(18 hours)**

Sadat Hassan Manto: The Dog of Tetwal

Intizar Hussain: A Chronicle of Peacocks

Selina Hossain: Fugitive Colours

Punakante Wijenaike: That Deep Silence

Module 3 (Life Writings) **(18 hours)**

Malcolm X: ―Nightmare, excerpt from *The Autobiography of Malcolm X*.

Sashi Deshpande: Learning to be a Mother in *Janani– Mothers, Daughters, Motherhood*, (Ed.) Rinki Bhattacharya.

Module 4 (Indigenous Identities **(18 hours)**

Leslie Marmon Silko: Lullaby

Garhwali Songs in *Painted Words- An Anthology of Tribal Literature* – Edited by G.N. Devy

Mamang Dai: Pinyar the Widow (Excerpt from Legends of Pensam)

Module 5 (Alter Identities) **(18 hours)**

Nathaniel Hawthorne: The Birth Mark

Girish Karnad: Hayavadana (Excerpt)

Ruskin Bond: The Girl on the Train

സെമസ്റ്റർ : മൂന്ന്

കോമൺ കോഴ്സ് II മലയാളം
MA3A05B18- ദൃശ്യകലാസാഹിത്യം

Credits: 4

Total Lecture hours: 90

പഠനനേട്ടങ്ങൾ (Course Outcomes)

CO1:കേരളീയരംഗകലാപാരമ്പര്യവും സംസ്കാരപരിണാമവും ചർച്ചചെയ്യുക

CO2:ദൃശ്യകലാപഠനത്തിലൂടെ കേരളീയസംസ്കാരപരിണാമം, ചരിത്രം എന്നിവ അപഗ്രഥിക്കുക

CO3:കഥാപാത്രപഠനത്തിലൂടെ സമകാലികവിഷയങ്ങളെ വിലയിരുത്തുക

CO4: ഇതിവൃത്ത പഠനത്തിലൂടെ കഥാപാത്രങ്ങളെ വിമർശനാത്മകമായി നിരൂപണം ചെയ്യുക

CO5:സമകാലികസംഭവങ്ങളെ അടിസ്ഥാനമാക്കി നാടകം, ഹ്രസ്വചിത്രം എന്നിവ തയാറാക്കുക.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	2	3
CO2	1	1	1	2	3
CO3	1	1	1	2	3
CO4	1	1	1	2	3
CO5	1	1	1	2	3

പാഠഭാഗങ്ങൾ

ഖണ്ഡം ഒന്ന് - സംസ്കൃത നാടകം **20 മണിക്കൂർ.**

മലയാളശാകുന്തളം നാലാമങ്കം - എ. ആർ രാജ രാജ വർമ

ഖണ്ഡം രണ്ട് - ആട്ടക്കഥ **15 മണിക്കൂർ**

നളചരിതം (ഒന്നാം ദിവസം) - ഉണ്ണായി വാര്യർ (തുടക്കം മുതൽ ഹംസം നളനിലുള്ള പ്രണയം ഉറപ്പിക്കുന്നത് വരെ)

ഖണ്ഡം മൂന്ന് - തുള്ളൽ **15 മണിക്കൂർ**

കല്യാണസൗഗന്ധികം (ശീതങ്കൻ തുള്ളൽ) - കുഞ്ചൻ നമ്പ്യാർ - ഭീമൻറെ കദളീവന പ്രവേശം മുതൽ ശ്രീരാമ ദാസൻറെ വംശേ ജനിക്കയാൽ പാരം നിനക്കു മഹംഭാവമിങ്ങനെ' വരെ ഭാഗങ്ങൾ

ഖണ്ഡം നാല് - മലയാള നാടകം **20 മണിക്കൂർ**

1128 ൽ ക്രൈം 27 - സി. ജെ. തോമസ്

ഖണ്ഡം അഞ്ച്- സിനിമ **20 മണിക്കൂർ**

നിർമാല്യം തിരക്കഥ - എം. ടി . വാസുദേവൻ നായർ

SEMESTER III
COMMON COURSE II– HINDI
HN3AO5B18 - NAATAK AUR LAMBI KAVITHA

Credits – 4

Total Lecturer Hours - 90

Course Outcomes:

CO1: Summarise the poems and illustrate the socio-political and cultural concerns of the Author

CO2: Discuss the Authors contribution to Hindi Literature

CO3: Analyse the characterisation of the Drama Konark

CO4: Critique excerpts of the poems and Drama

CO5: Communicate in oral and written form of Hindi with competence.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	3	3
CO2	1	1	1	3	3
CO3	1	1	1	2	3
CO4	1	1	1	2	3
CO5	1	1	1	3	3

Syllabus Content

Module- I **22 Hours**

Syllabus- Konark Introduction & Act 1 (Jagdishchandra Mathur)

Module- II **24 Hours**

Syllabus- - Konark Act 2 & 3(Jagdishchandra Mathur)

Module- III **22 Hours**

Syllabus- Nagayi Mahura (Thrilochan)

Shahenshah Ki Neend (Umashankar Chaudhary)

Dhaaba- Nilesh Raghuvanshi

Module- IV **22 Hours**

Syllabus- Ithni Door Mat Bhyahna Baba- Nirmala Putul

Jawahar Tunnel – Agnishekhar

SEMESTER III

COMMON COURSE II

FR3A05B18- AN ADVANCED COURSE IN FRENCH - I

Credits: 4

Total Lecture Hours: 90

Course Outcomes:

CO1: Describe topics such as physical appearance of a person, sports and entertainments.

CO2: Articulate the concepts to express one's opinion in a specific situation.

CO3: Compose conversations based on scenarios which help while shopping.

CO4: Articulate the concepts to give advice and instructions and to invite a person in a specific situation.

CO5: Construct conversations based on scenarios which help during medical and health consultations.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	1	1	3	2
CO2	1	1	1	3	2
CO3	1	1	1	3	2
CO4	1	1	1	2	3
CO5	1	1	1	3	2

Syllabus Content:

Module I **(30 hours)**

Jeunes artistes: Décrire une personne - Exprimer une opinion - La description physique - Les spectacles

Module II **(30 hours)**

Tenue de soirée : Inviter - Les vêtements - Les chaussures - Les couleurs - Les matières

Module III **(30 hours)**

Faites du sport ! : Donner des conseils - Les parties du corps - Les mouvements - Les sports

SEMESTER III

COMPLEMENTARY COURSE I

MT3C01B18-VECTOR CALCULUS, DIFFERENTIAL EQUATIONS AND ANALYTIC GEOMETRY

Credits: 4

Total Lecture Hours: 90

Course Outcomes:

CO1: Analyze the path, velocity and acceleration of moving bodies using Vector Calculus

CO2: Apply line and surface integrals to calculate Circulation, flux and work done by vector fields and potential function of conservative fields.

CO3: Apply Green's, Stokes and Divergence theorems to calculate multiple integrals.

CO4: Explain different types of differential equations and solve first order differential equations.

CO5: Classify conic sections and deduce their equations and geometric properties

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	1	3	1	1	2
CO2	1	3	1	1	2
CO3	1	3	1	1	2
CO4	1	3	1	1	2
CO5	1	3	1	1	2

Syllabus Content:

Module 1

Vector valued Functions: (15 Hrs)

Vector Functions, Arc length and unit Tangent vector **T**, Curvature and unit Normal Vector **N**, Torsion and unit Binormal vector **B,** Directional Derivatives and Gradient Vectors**. (Sections** 13.1, 13.3, 13.4, 13.5 and 14.5 of text 2)

Module 2

Integration in Vector Fields: (25 Hrs)

Line Integrals, Vector fields and Work, Circulation and Flux, Path independence, Potential Function and Conservation Fields, Green's theorem in Plane (Statement and problems only), Surface area and Surface integral, Parameterised Surface, Stoke'stheorem(Statement and Problems only), the Divergence theorem and a Unified theory (Statement and simple problems only). (Sections 16.1 to 16.8 of text 2)

Module 3

Ordinary differential equations: (25Hrs)

Exact Differential Equation, Linear Equations , Solutions by Substitutions, Equations of first order and not of first degree , First order equations of higher Degree solvable for p , Equations solvable for y , Equations solvable for x, Equations of first degree in x and y , Lagrange's and Clairaut's Equation (sections 2.1 , 2.2 , 2.3 , 2.4 , 3.1 , 3.2 , 3.3 , 3.4 , 3.5 of text 1)

Module 4

Analytic Geometry: (25Hrs)

Conic sections and Quadratic equations, Classifying Conic Sections by Eccentricity, Conics and Parametric equations, The Cycloid, polar co-ordinates, Conic Sections in Polar coordinates. (Sections 10.1, 10.2, 10.4, 10.5, 10.8 of Text 2) (exclude the pedal Method and Newtonian Method)

SEMESTER III

COMPLEMENTARY COURSE II

PH3C02B18 - MODERN PHYSICS AND BASIC ELECTRONICS

Credits: 3

Total lecture hours - 54 hrs

Course Outcomes:

CO1: Discuss different atom models used to study spectroscopy and estimate the spectral characteristics

CO2: Discuss emergence of quantum mechanics and solve photoelectric equation, energy and uncertainties in position/momentum of a particle in a box

CO3: Construct rectifiers, voltage regulators and explain the characteristics of transistors

CO4: Explain the ground state properties of the nucleus and demonstrate different types of nuclear disintegrations and combination

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	2	1	2
CO2	3	3	2	1	2
CO3	2	3	3	1	2
CO4	2	3	1	1	2

Syllabus Content:

Module I **(16 hours)**

Atom models & Spectroscopy

Thomson's model - Rutherford's nuclear atom model (qualitative) - Bohr atom model – Bohr radius – total energy of the electron – Bohr's interpretation of Hydrogen atom- Sommerfeld's

relativistic atom model – elliptical orbits of Hydrogen (qualitative) – Sommerfeld's relativistic theory – fine structure of Hα line - Vector atom model – quantum numbers associated with vector atom model – coupling scheme (qualitative) - optical spectra – spectral terms – spectral notation – selection rules.

Molecular spectra – theory of origin of pure rotational spectra of rigid diatomic molecule - Raman effect – experimental study of Raman effect – quantum theory of Raman effect-- fluorescence and phosphorescence.

Module II **(12 hours)**

Quantum mechanics

Introduction – breakdown of classical physics – black body radiation and Planck's quantum hypothesis (qualitative) – photoelectric effect – Einstein's explanation of photoelectric effect – de Broglie hypothesis – matter wave – Davisson Germer experiment – uncertainty principle (derivation and application not required) - wave packet – wave function – properties of wave function – probabilistic interpretation of wave function – normalisation condition – time independent Schrödinger equation – particle in a box problem.

Module III **(11 hours)**

Basic Electronics

Energy bands in solids - conduction in solids – semiconductors - majority and minority charge carriers - intrinsic conduction. PN junction diodes – biasing - diode equation (derivation not required), diode parameters, diode ratings - diode characteristics – junction break down. Rectifiers - half wave, full wave and bridge rectifiers. Zener diode characteristics – voltage regulation. Bipolar junction transistors – biasing - transistor currents - transistor circuit configurations - common emitter configurations.

Module IV **(15 hours)**

Nuclear Physics

Classification of nuclei - general properties of nucleus - binding energy - nuclear stability - theories of nuclear composition - nuclear forces - magic numbers - natural radioactivity - alpha- beta & gamma rays - properties of alpha rays - properties of beta rays - properties of gamma rays-

fundamental laws of radioactivity – Soddy Fajan's displacement law - law of radioactive disintegration – half life - mean life - units of radioactivity - law of successive disintegration - radioactive dating.

Nuclear Fission & Fusion **(7 hrs)**

Nuclear fission- Energy released in fission- Chain reaction- Atom bomb- Nuclear reactors - Nuclear Fusion- Source of stellar energy- Thermonuclear reactions- Transuranic elements

SEMESTER III

CORE COURSE

CH3B03B18: ORGANIC CHEMISTRY – I

Credits – 3

Total Lecture Hours: 54

Course Outcomes:

CO1: Summarize the types of reagents and reactive intermediates, electronic displacement effects and reaction mechanisms in organic chemistry.

CO2: Apply the knowledge of isomerism in predicting the nomenclature and stability of organic molecules.

CO3: Generalize the preparation, properties and uses of alkanes, alkenes, alkynes, alkyl halides and aryl halides.

CO4: Predict the aromaticity and reactivity of aromatic compounds towards Electrophilic substitution reactions.

Mapping of Course Outcomes with Program Specific Outcomes

Mapping	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	2	2	1	1
CO2	3	3	2	1	2
CO3	3	2	2	1	1
CO4	3	3	2	1	2

Syllabus Content:

Module 1: Fundamentals of Organic Chemistry (12 Hrs)

Classification and IUPAC system of nomenclature of common organic compounds (both aliphatic and aromatic). Line diagram drawing. Factors affecting reaction mechanism. Polarity of bonds.

Electronic displacements: Inductive effect, electromeric effect, mesomeric effect, resonance and hyperconjugation. steric effects.

Cleavage of bonds: Homolysis and heterolysis with suitable examples. curly arrow rules, formal charges. *Types of reagents*: Nucleophiles and electrophiles. *Reactive intermediates*: Carbocations, carbanions, free radicals and carbenes – types, shape and relative stability. *Types of organic reactions*: Addition, elimination, substitution, rearrangement and redox reactions (definition and one example each). *Pericyclic Reactions*: Classification – electrocyclic reactions, cycloadditions - Diels-Alder reaction and Sigmatropic rearrangements - Claisen rearrangement (with mechanism).

Module II : Stereochemistry (15 Hrs)

Stereoisomerism – definition, classification.

Optical isomerism: Optical activity, specific rotation, concept of chirality (upto two carbon atoms). Configuration. Enantiomerism, diastereomerism and meso compounds. Racemic mixture and methods of resolution. Asymmetric synthesis (partial and absolute). Threo and erythro; *d* and *l* designations; Cahn-Ingold-Prelog rules: R/ S notation (for upto 2 chiral carbon atoms).

Geometrical isomerism: cis–trans, *syn-anti* and E/Z nomenclature (for upto two C=C systems) with C.I.P rules. Methods of distinguishing geometrical isomers.

Conformational analysis: Conformational analysis with respect to ethane, butane and cyclohexane. Relative stability and energy diagrams. Interconversion of Wedge formula, Newman, Sawhorse and Fischer projection formulae. Chair, boat and twist boat forms of cyclohexane with energy diagrams. Conformation of methyl cyclohexane. Origin of ring strain in cyclic systems. Baeyer's strain theory.

Module III: Aliphatic Hydrocarbons and Alkyl Halides (12 Hrs)

Alkanes: Preparation - catalytic hydrogenation, Wurtz reaction, Wurtz-Fittig reaction, from Grignard reagent. Reactions - free radical substitution - halogenation. *Alkenes:* Preparation - Elimination reactions - mechanism of E1 and E2 reactions. Dehydration of alkenes and dehydrohalogenation of alkyl halides (Saytzeff's and Hofmann's rules). Reactions - *cis*-addition

(alkaline $KMnO_4$) and *trans*-addition (bromine). Addition of HX (Markownikoff's and anti-Markownikoff's addition with mechanisms), Hydration, Ozonolysis.

Alkynes: Preparation - Acetylene from CaC_2 and conversion into higher alkynes; by dehalogenation of tetra halides and dehydrohalogenation of vicinal dihalides. Reactions - Acidity of alkynes, formation of metal acetylides, alkylation of terminal alkynes and conversion into higher alkynes, addition of bromine and alkaline $KMnO_4$.

Alkyl Halides: Preparation - From alkenes and alcohols. Reactions - Types of aliphatic nucleophilic substitution reactions - $S_N1$and S_N2 mechanisms with stereochemical aspects and effects of substrate structure, solvent, nucleophile and leaving group.

Organometallic compounds of Mg (*Grignard reagents*) – Formation, structure and important reactions/synthetic applications.

Module IV: Aromatic Hydrocarbons and Aryl Halides (15 Hrs)

Aromaticity : Definition, Hückel's rule - application to benzenoid (benzene, naphthalene and anthracene) and non-benzenoid (cyclopropenyl cation, cyclopentadienyl anion and tropylium cation) compounds.

Benzene: Molecular orbital picture and resonance energy. Preparation - from phenol, by decarboxylation, from acetylene, from aromatic acids. Reactions - Electrophilic aromatic substitution: nitration, halogenation, sulphonation and Friedel-Craft's reaction (alkylation and acylation) with their mechanism.

Orientation of aromatic substitution. *ortho, para* and *meta* directing effects of groups. Ring activating and deactivating groups with examples.

Naphthalene and Anthracene: Molecular orbital picture and resonance energy. Preparation (of Naphthalene): Haworth synthesis, Reactions - Electrophilic substitutions (halogenation, nitration and sulphonation) of naphthalene.

Aryl Halides: Preparation - chloro, bromo and iodo-benzene from phenol, Sandmeyer and Gattermann reactions. Reactions - aromatic nucleophilic substitutions – bimolecular displacement mechanism, elimination-addition (benzyne intermediate) mechanism.